

Ammonium analysis

Based on:

- Weatherburn, M. W. 1967. Phenol-hypochlorite reaction for determination of ammonia. *Analytical Chemistry* 39:971-974.
- Steve Allison, Allison Lab Protocol: Nutrient Analysis, 1/2008
- BOWER, . E., AND T. HOLM-HANSEN 1980. A salicylate-hypochlorite method for determining ammonia in seawater, *Can. J. Fish. Aquat. Sci.* 37: 794-798.

The salicylate method is a variation of the Berthelot-Phenate method but does not require the use and disposal of toxic phenol. The salicylate method involves a three-step reaction sequence. The first reaction step involves the conversion of ammonia to monochloramine by the addition of chlorine. The monochloramine then reacts with salicylate to form 5-aminosalicylate. Finally, the 5-aminosalicylate is oxidized in the presence of sodium nitroferricyanide (a catalyst) to form a blue-green colored dye that absorbs light at 650nm

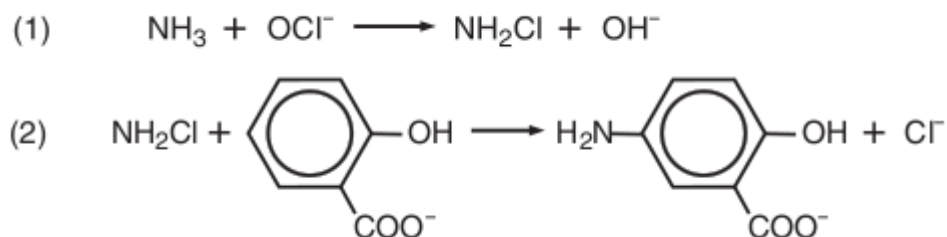


Fig 1: Ammonia compounds are initially combined with hypochlorite to form monochloramine (1), which then reacts with salicylate to form 5-aminosalicylate (2).

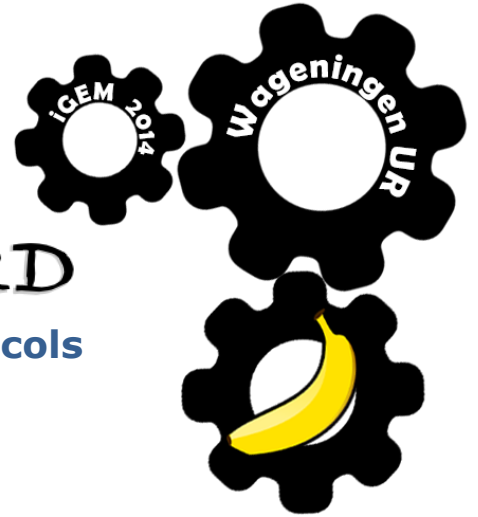
Note: This reaction is light sensitive. Perform in a dark environment (<85 lux) or add to every sample a calibration curve to counteract the degradation caused by light. For the same reason it is important to use a multichannel pipet to start every reaction at the same time.





GUARD

Protocols



Stock ammonium solution:

- 0.23585 g Ammonium sulphate/chloride
- 500 ml ultrapure water.

Sodium salicylate solution

- 6.8 g sodium salicylate
- 5.6 g Trisodium Citrate, Dihydrate
- 5 g Sodium Potassium tartrate, Tetrahydrate
- 0.025 g sodium nitroprusside (Disodium Pentacyanonitrosylferrate, Dihydrate)
- 100 ml ultrapure water

Sodium hydroxide solution

- 6 g sodium hydroxide
- 100 ml ultrapure water

Bleach solution (make fresh each day)

- 0.2 ml bleach (sodium hypochlorite)
- 9.8 ml sodium hydroxide solution

Standard curves: Dilute the 100 ppm stock solution to 10 ppm in a 1.5 ml centrifuge tube (150 μ l

stock:1350 μ l ultrapure water). Create the following standard curves in 1.5 ml centrifuge tubes.

High concentration (0-10 ppm)			Low concentration (0-1 ppm)		
Concentration	μ l 10 ppm mix	μ l water	Concentration	μ l 10 ppm mix	μ l water
0 ppm	0	1000	0 ppm	0	1000
0.5	50	950	0.05	50	950
1.0	100	900	0.10	100	900
2.0	200	800	0.20	200	800
5.0	500	500	0.50	500	500
10.0	1000	0	1.00	1000	0

Detection limit <0.05 ppm

For 96-well plate (~200 μ l):

For low concentrations (0-5 ppm):

Add the following to each well:

1. 80 μ l sample
2. 60 μ l bleach solution (add using multichannel pipet)
3. 60 μ l salicylate solution (add using multichannel pipet)

For high concentrations (1-10 ppm):

1. 20 μ l sample
2. 90 μ l bleach solution (add using multichannel pipet)
3. 90 μ l salicylate solution (add using multichannel pipet)

Pipet up and down to mix well, incubate 50 min and read plate at 650 nm.

